

Monday 21 May 2012 – Morning

**GCSE GATEWAY SCIENCE
ADDITIONAL SCIENCE B**

B624/01 Unit 2 Modules B4 C4 P4 (Foundation Tier)



Candidates answer on the Question Paper.
A calculator may be used for this paper.

OCR supplied materials:

None

Other materials required:

- Pencil
- Ruler (cm/mm)

Duration: 1 hour



Candidate forename					Candidate surname				
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Centre number						Candidate number			
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INSTRUCTIONS TO CANDIDATES

- Write your name, centre number and candidate number in the boxes above. Please write clearly and in capital letters.
- Use black ink. HB pencil may be used for graphs and diagrams only.
- Answer **all** the questions.
- Read each question carefully. Make sure you know what you have to do before starting your answer.
- Write your answer to each question in the space provided. Additional paper may be used if necessary but you must clearly show your candidate number, centre number and question number(s).
- Do **not** write in the bar codes.

INFORMATION FOR CANDIDATES

- The number of marks is given in brackets [] at the end of each question or part question.
- A list of physics equations is printed on page two.
- The Periodic Table is printed on the back page.
- The total number of marks for this paper is **60**.
- This document consists of **24** pages. Any blank pages are indicated.

EQUATIONS

$$\text{speed} = \frac{\text{distance}}{\text{time taken}}$$

$$\text{acceleration} = \frac{\text{change in speed}}{\text{time taken}}$$

$$\text{force} = \text{mass} \times \text{acceleration}$$

$$\text{work done} = \text{force} \times \text{distance}$$

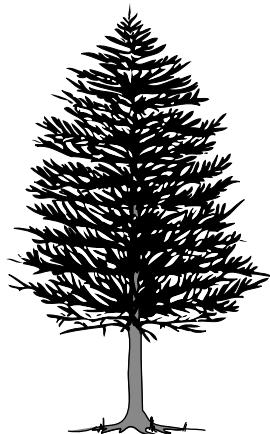
$$\text{power} = \frac{\text{work done}}{\text{time}}$$

$$\text{resistance} = \frac{\text{voltage}}{\text{current}}$$

Answer **all** the questions.

Section A – Module B4

- 1 Large areas of Canada are covered by forests containing pine trees.



- (a) Pine trees make food by photosynthesis.

- (i) Which part of the pine tree has the most cells that photosynthesise?

..... [1]

- (ii) Small green structures inside plant cells absorb light energy for photosynthesis.

Write down the name of these structures.

..... [1]

- (iii) For photosynthesis, the pine trees need water and carbon dioxide.

Put ticks (\checkmark) in **two** boxes to show where these substances enter the tree.

water enters through leaf pores

carbon dioxide enters through leaf pores

water enters through roots

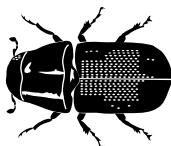
carbon dioxide enters through roots

water enters through flowers

carbon dioxide enters through flowers

[2]

- (b) Many of the pine trees in Canada are being killed by small beetles.



The mountain pine beetle feeds from the tubes in the tree that transport sugar.

The beetle also infects the tree with a fungus.

This fungus blocks up the xylem vessels in the tree trunk.

This kills the tree.

- (i) Write down the name of the tubes from which the beetle feeds.

..... [1]

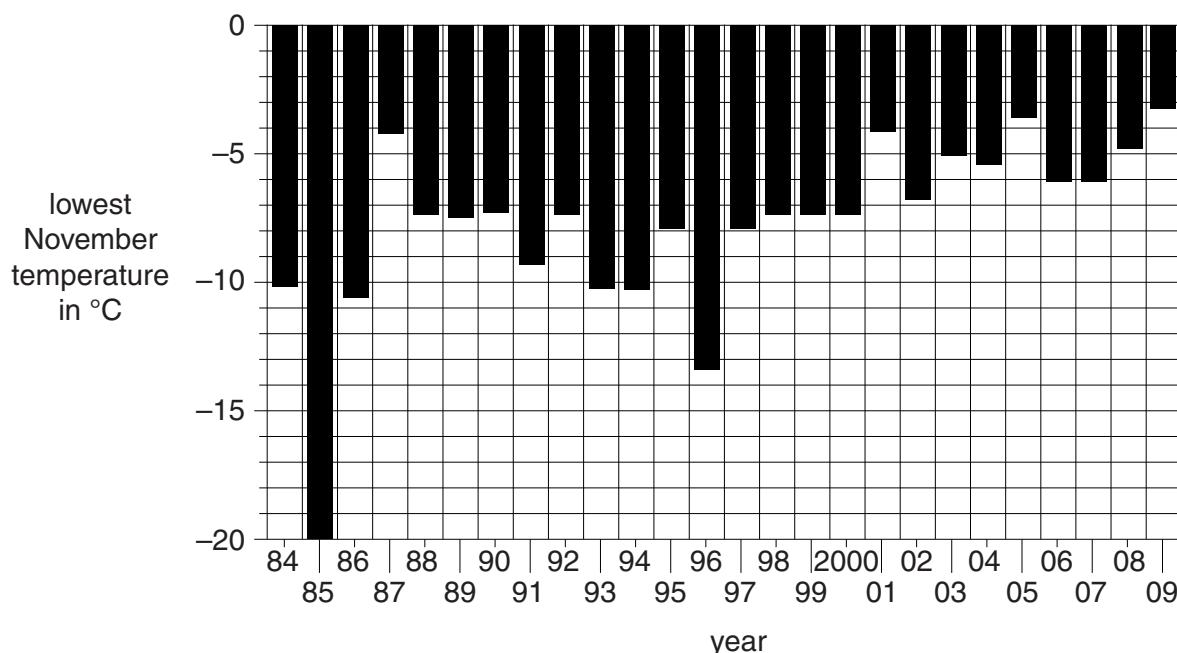
- (ii) The dead trees are used for fuel and burned.

Write down **one other** type of biomass that is burned for fuel.

..... [1]

- (c) Many of the beetles are killed if temperatures drop below -6°C in November.

The graph shows the lowest November temperature in the pine forest each year from 1984 to 2009.



- (i) In 1988 there was a large outbreak of the beetle but they soon decreased in number.

Use the graph to work out in which year the next major outbreak started.

..... [1]

- (ii) The recent outbreak has lasted for a number of years.

Scientists are worried that it might continue for many more years.

Use the graph to suggest why.

.....
.....
..... [2]

- (d) The scientists are planning to use biological control to save the pine trees.

They are releasing another type of beetle called the clerid beetle into the forests.

Suggest how this beetle acts as a biological control.

..... [1]

[Total: 10]

2 Bill grows lettuces on his farm.

(a) He finds that the lettuces are being eaten by slugs and birds.

The slugs are being eaten by hedgehogs.

(i) The lettuces are **producers** in this feeding relationship.

What is meant by the word producer?

..... [1]

(ii) Sketch a **pyramid of biomass** to represent these feeding relationships.

Write the names of the organisms in the correct boxes in the pyramid.

[2]

(b) Bill investigates a new way of growing lettuces.

This involves growing them in tanks without soil.

Water containing minerals is pumped through the tanks.

What name is given to this method of growing lettuces without soil?

Put a **ring** around the correct answer in this list.

battery farming

crop rotation

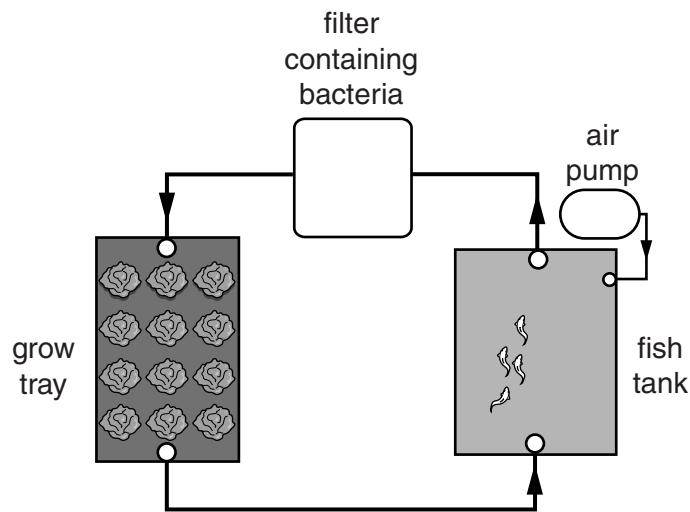
hydroponics

transpiration [1]

- (c) The diagram shows the equipment that Bill uses.

He keeps fish in a tank and then circulates the water through a filter containing bacteria.

The bacteria will convert waste made by the fish into minerals needed by the lettuce.



- (i) Which part of the lettuce plant takes up the minerals?

..... [1]

- (ii) Bill adds a chemical to the water that will kill pests.

Suggest **one** reason why he wants to kill pests.

..... [1]

[Total: 6]

- 3 Read this information about preserving food.

In the 18th century, Napoleon was travelling in hot countries with his army.

The hot conditions meant that the food often started to rot.



The French government offered a large sum of money to anybody who could find a way to stop the food rotting.

Nicolas Appert solved the problem.

He put food into glass bottles and used steam to heat the bottles to high temperatures.

He then removed the steam and quickly sealed each bottle with a cork and wax.

Appert won the prize and set up his own factory.

- (a) Write down **one** type of microorganism that was responsible for rotting Napoleon's food.

..... [1]

- (b) Explain why Appert needed to seal each bottle after heating it with steam.

..... [1]

- (c) The principle used in Appert's method is still used in one type of food preservation.

Draw a straight line to join each **type of food preservation** to the **principle** that it uses.

type of food preservation

drying

principle

Appert's method

canning

adding vinegar

pickling

removing water

[2]

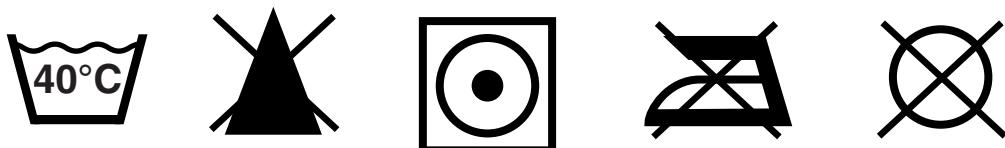
[Total: 4]

Section B – Module C4

- 4 Miles is doing the washing.

He washes a shirt.

Look at the wash label on his shirt.



- (a) What do these symbols tell you about how Miles should wash, dry and iron his shirt?

.....
.....
.....
.....

[3]

- (b) Miles washes his shirt at a low temperature.

This saves him money.

Write down one **other advantage** of washing clothes at lower temperatures.

..... [1]

- (c) Miles uses a biological washing powder that contains **enzymes**.

What is the job of the enzymes in the washing powder?

.....
.....

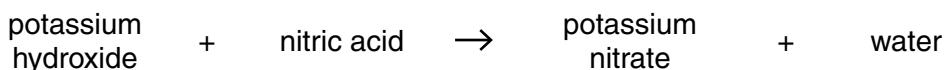
[1]

[Total: 5]

- 5 Kylie makes a fertiliser called potassium nitrate.

She adds an alkali called potassium hydroxide to nitric acid.

Look at the word equation.



- (a) What is the name of this **type** of reaction?

Choose from this list.

catalysis

neutralisation

precipitation

reversible

answer

[1]

- (b) Kylie uses 5.6g of potassium hydroxide and 6.3g of nitric acid.

She makes 10.1g of potassium nitrate.



- (i) What mass of water does she make?

answer g

[1]

- (ii) Kylie does the experiment again.

This time she uses 16.8g of potassium hydroxide and 18.9g of nitric acid.

Calculate the mass of potassium nitrate she will make.

.....
.....

answer g

[1]

- (iii) Kylie does the experiment a third time.

She expects to make 5.5 g of potassium nitrate.

She actually makes 4.4 g.

Show by calculation that her percentage yield is 80%.

.....
.....
.....

[2]

- (c) The formula for potassium nitrate is KNO_3 .

Which essential element for plant growth is **not** in potassium nitrate?

..... [1]

- (d) Farmers add fertilisers to their fields.

This is so they can make more money from their crops.

Write about why adding fertilisers will make more money for the farmer.

.....
.....
.....

[2]

[Total: 8]

12

- 6 Carbon exists in three solid forms at room temperature.

These are diamond, graphite and buckminsterfullerene.

- (a) Match each **form of carbon** with its **properties**.

Draw only three straight lines.

form of carbon	properties
diamond	hard with a high melting point
graphite	black solid which dissolves in petrol to make a red solution
buckminsterfullerene	conducts electricity

[2]

- (b) Diamond is used for making jewellery.

Write down **one other** use of diamond.

..... [1]

- (c) Graphite is used to make pencil leads.

Explain why.

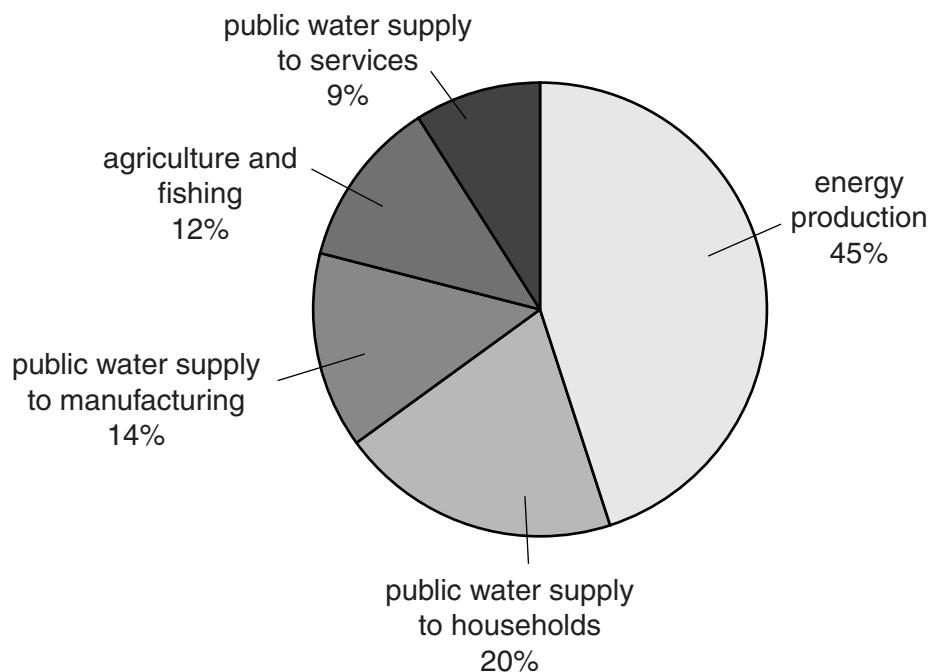
.....

.....

..... [2]

[Total: 5]

- 7 Look at the pie chart. It shows the total use of water in the UK in 2005.



- (a) Which sector used the **greatest** percentage of water?

..... [1]

- (b) What is the **total** percentage of water used for the public water supply?

..... [1]

[Total: 2]

Section C – Module P4

- 8 This question is about static electricity.

Danni combs her hair.



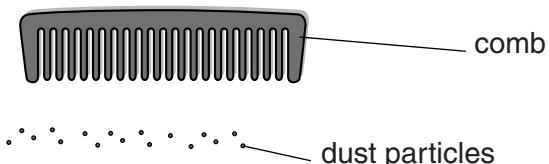
- (a) Both the comb and her hair become **charged**.

Complete this sentence.

If Danni's hair has a charge then the comb has
a charge. [2]

- (b) She places the comb near some dust particles.

The dust particles are **uncharged**.



- (i) What happens to the dust particles?

.....

..... [1]

- (ii) Describe how this idea can be useful in the **home**.

.....

..... [1]

(c) Write about two **other** uses of static electricity.

1

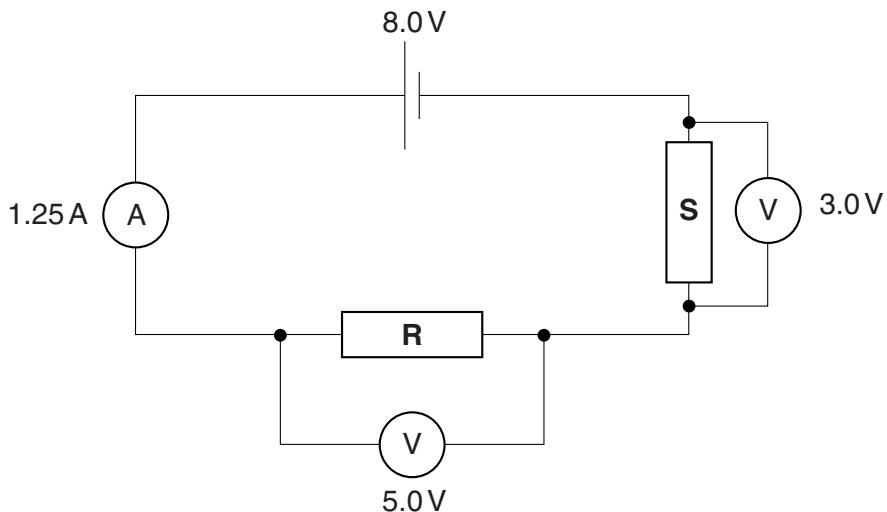
2

[2]

[Total: 6]

- 9 Ramiz is investigating electrical circuits.

He builds a circuit with two resistors in it.



- (a) Ramiz measures

- the current in the circuit
- the voltage across each resistor.

Calculate the **resistance** of resistor **R**.

The equations on page 2 may help you.

.....

resistance of **R** = ohms (Ω)

[2]

- (b) (i) Ramiz adds another identical cell to his circuit.

This doubles the supply voltage.

This increases the current flowing through the resistor.

What happens to the **value** of the **voltage** across resistor **R**?

.....

[1]

- (ii) He adds a third resistor to the circuit.

It is placed in series with the other resistors.

What happens to the current in the circuit?

.....

..... [1]

[Total: 4]

10 Radioactive isotopes emit nuclear radiation.

Nuclear radiation has many uses in hospitals.

One use is as a **tracer**.



(a) Write about the **other** uses of nuclear radiation in hospitals.

In your answer, include the

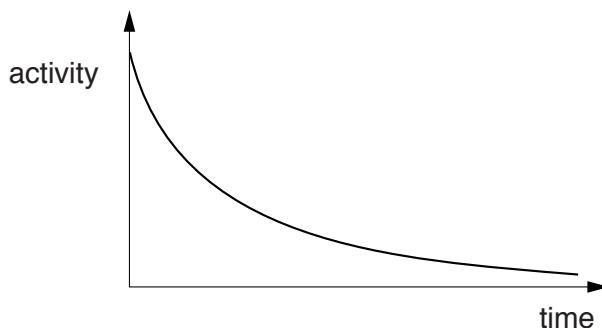
- name given to the medical **person** using the radiation
- **other** uses of radioactive isotopes in hospitals
- **type** of radiation used.

[3]

- (b) When a tracer is put into a patient it is tracked around the body.

- (i) Tracers are radioactive.

The graph shows how the activity (radioactivity) of a tracer changes over time.



Use the graph to describe what happens to the activity of the tracer over time.

.....
.....

[1]

- (ii) Nuclear radiation comes from part of an unstable atom.

Write down the name of this **part** of the atom.

.....

[1]

- (c) Complete the sentences using phrases from this list.

a beta particle

a gamma ray

an X-ray

an alpha particle

An unstable atom emits a helium nucleus.

Scientists call this helium nucleus

An unstable atom emits a fast moving electron.

Scientists call this fast moving electron

[2]

[Total: 7]

11 Abbie finds out that **ultrasound** is used in hospitals.

(a) Look at the statements about ultrasound.

Put ticks (\checkmark) in the boxes beside the **three** correct statements.

it is used to measure blood flow

it is used to sterilise hospital equipment

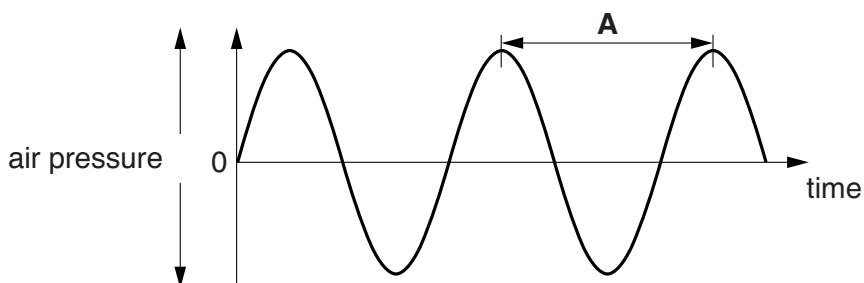
it is a longitudinal wave

it is used to break up kidney stones

it is an electromagnetic wave

[2]

(b) Ultrasound waves can be shown as a wave diagram.



What feature of the wave is shown by the letter **A**?

..... [1]

[Total: 3]

END OF QUESTION PAPER

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The Periodic Table of the Elements

1 2

1	H	hydrogen	1
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Key			
relative atomic mass atomic symbol name	atomic (proton) number		

7	Li	lithium	3
9	Be	beryllium	4

23	Na	sodium	11
24	Mg	magnesium	12

39	40	45	48	51	52	55	56	59	63.5	65	70	73	75	79	80	84
K		Sc	Ti	V	Cr	Mn	Fe	Co	copper	Zn	Ga	Ge	As	Se	Br	Kr
potassium	19	scandium	titanium	vanadium	chromium	manganese	iron	nickel	nickel	zinc	gallium	germanium	arsenic	selenium	bromine	krypton
		21	22	23	24	25	26	27	28	30	31	32	33	34	35	36
85	88	89	91	93	96	[98]	101	103	Ru	Pd	Cd	In	Sb	Te	I	Xe
Rb		Sr	Y	Nb	Mo	Tc	Ru	Rh	ruthenium	palladium	cadmium	indium	antimony	tellurium	iodine	xenon
rubidium	37	strontium	yttrium	niobium	molybdenum	technetium	43	45	44	46	48	49	51	52	53	54
		38	39	40	41											
133	137	139	178	181	184	186	190	192	Ir	Pt	Tl	Bi	Po	[209]	[210]	[222]
Cs		La*	Hf	Ta	W	Re	Os	Iridium	iridium	platinum	mercury	thallium	bismuth	polonium	astatine	Rn
caesium	55	lanthanum	hafnium	tantalum	tungsten	rhenium	osmium	77	77	78	80	81	83	84	85	radon
		57	72	73	74	75	76									
[223]	[226]	[227]	[261]	[262]	[266]	[264]	[268]	[277]	Mt	Ds	[271]	[272]	Rg	[209]	[210]	[222]
Fr		Ac*	Rf	Db	Sg	Bh	Hs		meitnerium	darmstadtium						
francium	87	actinium	rutherfordium	dubnium	seaborgium	bohrium	hassium	109	metastable	110						
		89	104	105	106	107	108		109							

Elements with atomic numbers 112-116 have been reported but not fully authenticated

1	2	3	4	5	6	7	0	4
								He

24

* The lanthanoids (atomic numbers 58-71) and the actinoids (atomic numbers 90-103) have been omitted.

The relative atomic masses of copper and chlorine have not been rounded to the nearest whole number.