

# Cambridge Technicals Applied Science

### **Unit 3: Scientific Analysis and Reporting**

Level 3 Cambridge Technical in Applied Science 05847 – 05849/05874/05879

# Mark Scheme for January 2022

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This mark scheme is published as an aid to teachers and students, to indicate the requirements of the examination. It shows the basis on which marks were awarded by examiners. It does not indicate the details of the discussions which took place at an examiners' meeting before marking commenced.

All examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes should be read in conjunction with the published question papers and the report on the examination.

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Annotations available in RM Assessor

Annotation	Meaning
$\checkmark$	Correct response
×	Incorrect response
<b>^</b>	Omission mark
BOD	Benefit of doubt given
CON	Contradiction
RE	Rounding error
SF	Error in number of significant figures
ECF	Error carried forward
L1	Level 1
L2	Level 2
L3	Level 3
NBOD	Benefit of doubt not given
SEEN	Noted but no credit given
I	Ignore

#### **Mark Scheme**

#### Subject-specific Marking Instructions

#### INTRODUCTION

Your first task as an Examiner is to become thoroughly familiar with the material on which the examination depends. This material includes:

- the specification, especially the assessment objectives
- the question paper
- the mark scheme.

You should ensure that you have copies of these materials.

You should ensure also that you are familiar with the administrative procedures related to the marking process. These are set out in the OCR booklet **Instructions for Examiners**. If you are examining for the first time, please read carefully **Appendix 5 Introduction to Script Marking: Notes for New Examiners**.

Please ask for help or guidance whenever you need it. Your first point of contact is your Team Leader.

Q	Question		Answer		Guidance
1	1 (a)		Laboratory Technology Reports ✓		
	(b)	(i)	2016 🗸	1	
	(ii) 3 (months) √		1		
	(c) Two different addresses/institutions/countries (in the affiliations) / (some authors are in) Germany and (others in) Spain √		1	ALLOW reference to groups a and b	
	(d) Germany √		1		
	(e)	(e) The paper was available online <b>2 days</b> after it was accepted $\checkmark$		1	ALLOW reference to short time period
	(f)	(f) (i) Any two from: Review by other scientists/researchers/experts/teams ✓ Working in the same field ✓ Prior to publication ✓		2	<b>IGNORE</b> 'others' <b>DO NOT ALLOW</b> friends / family (negates the first marking point)
(ii) Any one from: The findings are chee Ensures the quality/va		(ii)	Any one from: The findings are checked (before publication) ✓ Ensures the quality/validity of published research/data ✓	1	IGNORE to check for errors/mistakes ALLOW AW for data e.g., results / conclusions / methods
			Total	9	

Question		on	Answer		Guidance
2	2 (a)		all bars drawn to correct height (and same width) $\checkmark$		CHECK every bar
			both axes with appropriate labels $\checkmark$		ALLOW e.g., percentage / % on y-axis, colour on x-axis
			gaps between bars (of equal distance) $\checkmark$		
	(b)	(i)	Vertical lines at 450 and 700 nm $\checkmark$	1	
		(ii)	A single colour is a range of wavelengths       ✓         Colour is discontinuous but wavelength is continuous       ✓         Colour is continuous but wavelength is discontinuous       ✓         Stars emit a range of wavelengths       ✓         The wavelength ranges are irregular       ✓	2	
		(iii)	550 (x 10-9 m) √		ALLOW 1 mark for any correct wavelength colour pair
		(iv)	(It) appears white because it emits the spectrum / full range of visible light $\checkmark$	1	ALLOW (it emits) a mixture of colours
	(c)	<ul> <li>(i) M-type stars are the coolest type of star√ AND M-type stars emit wavelengths which are in the red colour range so appear red √</li> <li>OR K-type stars are cooler than G-type stars √ AND K-type stars emit wavelengths which are at the end of the red colour range so appear red √</li> </ul>		2	<b>ALLOW</b> e.g., K-stars have more (a peak of) red wavelengths at 4000 °C but G-stars have more (a peak of) blue wavelengths at 5000 °C (= 2 marks)

Mark Scheme

Qı	uestion	Answer	Mark	Guidance
	(ii)	<b>Correct</b> because: Most/51% of the stars (in <b>Fig. 2.1</b> ) are blue $\checkmark$ ( <b>Fig. 2.3</b> trend shows) blue is hotter $\checkmark$ <b>Incorrect</b> because: these are the brightest stars (seen from Earth) not the nearest stars (in <b>Fig. 2.1</b> ) $\checkmark$ there may be nearer stars that are too dim to see/dimmer/not included in data/do not emit visible light	4	ALLOW only 40% / fewer are yellow or red
		Total	15	

C	Question		Answer	Mark	Guidance
3	(a)	(i)	1.6 <b>OR</b> 1.60 ✓	2	
			1.58 ✓		
		(ii)	FIRST CHECK ANSWER ON ANSWER LINE If answer = 1.55 (V) award 3 marks		
			(M1) 10.87 ✓		
			(M2) 10.87 ÷ 7 = 1.5528 ✓		ALLOW ECF using their value from M1
			(M3) = 1.55 (3sf) ✓		ALLOW ECF using their value from M2 recorded to 3 s.f.
	(b)		FIRST CHECK ANSWER ON ANSWER LINE If answer for s = and s <sup>2</sup> = award 6 marks (M1) $(Xi - \overline{X}) : 0.05, -0.07, 0.02, 0.05, -0.11, 0.03 \checkmark$	6	
			(M2) $(X_i - X)^2$ : 0.0025, 0.0049, 0.0004, 0.0025, 0.0025, 0.0121, 0.0009 $\checkmark$ (M3) $\Sigma (X_i - \overline{X})^2 = 0.0258 \checkmark$		
			(M4) variance, s <sup>2</sup> , = 0.0258 ÷ 6 = 0.00429 $\checkmark$		ALLOW their M3÷6
			(M5) answer M4 recorded as $s^2 = 0.00429 \checkmark$		ALLOW their M3÷6 recorded as s <sup>2</sup>
			(M6) standard deviation, (s = $\sqrt{0.00429}$ ) = 0.06549 $\checkmark$		ALLOW their M5 correctly square-rooted ALLOW calculator value of s = 0.0655 / 0.066 / 0.07
	(c)	(i)	Any <b>two</b> batteries from 2, 3 or 6 $\checkmark$	1	
		(ii)	other values are similar to original measurements $\checkmark$	2	ALLOW these values are too different
			other values have same increase in precision / other values within 5% / these values more than 5% different $\checkmark$		ALLOW other values are within a narrower range

Mark Scheme

C	Question		Answer		Mark	Guidance
	(d)	on	term          accuracy         measurement error         precision         random error         systematic error	Answer         definition         the closeness of agreement between measured values obtained by repeated measurements         derror due to measurements varying in an unpredictable way         error due to measurements differing from the true value by a consistent amount         the closeness of the instrument reading to the true value         the difference between a measured value from the true value $\bigvee \bigvee $	5	Guidance
				Total	19	

Unit 3

C	Question		Answer		Guidance
4	(a)	(i)	h = 2.0  (cm) <b>AND</b> $t = 79  (s)$	1	
		(ii)	FIRST CHECK ANSWER ON ANSWER LINE If answer = 1.00 cm <sup>3</sup> s <sup>-1</sup> award 3 marks $[3.14 \times (7.1^2) \times 2.0] \div [4 \times 79] \checkmark$	3	ALLOW ecf from (a)(i)
			1.00 ✓ cm <sup>3</sup> s <sup>-1</sup> √		ALLOW answer that rounds to 1 (1 s.f.) ALLOW cm <sup>3</sup> /s, and other consistent units
		(iii)	( <i>t</i> <sub>2</sub> ) $\blacktriangle$ at 14, 330 (s) - using left hand scale $\checkmark$ (R) × at 14, 1.00 (cm <sup>3</sup> s <sup>-1</sup> ) - using right hand scale $\checkmark$	2	ALLOW ECF from (a)(ii)
		(iv)	Range of 3.1 to 3.2 $\checkmark$	1	IGNORE units
	(b)		[Level 3] Candidate shows a high level of understanding AND gives a detailed explanation of the evidence from Fig 4.2 and Fig 4.3. (5 - 6 marks) [Level 2] Candidate shows an understanding AND gives a simple explanation of the evidence from Fig 4.2 and Fig 4.3. (3 - 4 marks) [Level 1] Candidate shows a basic understanding AND/OR gives a basic explanation of the evidence from Fig 4.2 OR Fig 4.3. (1 - 2 marks) [Level 0] Candidate response includes fewer than two valid points. (0 marks)	6	<ul> <li>Trend in <i>R</i> (Fig. 4.2) <ul> <li><i>R</i> decreases as h increases (simple)</li> </ul> </li> <li><i>R</i> decreases as t increases (simple)</li> <li><i>R</i> tends to zero as the amount of water in the can decreases to zero (detailed)</li> </ul> <li>Suggested reasons for trend in <i>R</i> (Fig. 4.2) <ul> <li>amount of water in the can decreases (simple)</li> <li>weight / force of water in the can decreases (detailed)</li> </ul> </li> <li>water pressure at the hole decreases (detailed)</li> <li>Explanation for the inaccuracy in determining the time measurements (Fig. 4.3) <ul> <li>the water surface is uneven / has bubbles making it difficult to read the water level (simple)</li> </ul> </li>

Questio	n Answer	Mark	Guidance
			<ul> <li>at 14 cm, the rate is slow, so it is difficult to decide exactly when the water reaches the 14 cm mark / the water level changes little over a long period of time (detailed)</li> <li>reference to taking reading from the bottom of the meniscus / uncertainty in where the meniscus is</li> </ul>
(c)	Acceleration = change in speed ÷ time         Density = mass ÷ volume         Force = mass x acceleration         Pressure = density x gravitational field         strength x depth	1	
Total		14	

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C	Question		Answer	Mark	Guidance
5	Question 5 (a) - (b) (i)		AnswerUse ticks or crosses with subscript letters to indicate marking pointsy-axis labelled 'time' (s) $\checkmark_y$ x-axis labelled 'decrease in distance of pendulum bob' (cm) $\checkmark_x$ appropriate scale labelled at least every third large square $\checkmark_s$ all points correctly plotted (check and circle the plot furthest from the best fit line) $\checkmark_p$ outlier identified/circled at (4/200) $\checkmark_o$ appropriate curved line of best fit for 54 cm pendulum $\checkmark_l$ appropriate curved line of best fit for 130 cm pendulum disregards outlier $\checkmark_l$ 207 $\checkmark$ 253 $\checkmark$		Guidance Guidan
	(b) (i)		207 ✓ 253 ✓	2	<b>ALLOW</b> any equal tolerance either side of 230 for 1 mark e.g. 220 <b>and</b> 240
	(ii)		range bar at 6 cm <b>AND</b> $\pm$ 10% from 207 to 253 $\checkmark$	1	ALLOW ECF using (b)(i) values

### Mark Scheme

Question	Answer		Mark	Guidance
(iii)	A larger decrease in the distance from the vertical position with each swing A smaller decrease in the distance from the vertical position with each swing The pendulum bob changing direction more quickly The pendulum bob changing direction more slowly An increase in the time for one swing of the pendulum bob	✓ ✓	2	
		Total	12	

Un	it	3
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Question		Answer		Mark	Guidance		
6	(a)		Label 1 2 3 4 5 6 7 √√√√√	Identification          acute         obtuse <i>P. pulcher</i> acuminate <i>P.perfoliatus P.robbinsii P.crispus</i>		7	<b>ALLOW</b> species without the <i>P</i> . prefix, for labels 3, 5, 6 and 7
	(b)	(i) (ii)	(they) share many common features $\checkmark$ underwater leaf width has large overlap $\checkmark$		2	ALLOW same underwater leaf blade shape ALLOW same floating tip shape ALLOW the only difference is the shape of the underwater leaf tip	
		(11)	number	of petals / stem width / AVP $\checkmark$	iower colour /	1	NUT JUST e.g., colour / stem
	(c)		genus √ binomia	<ul><li>✓</li></ul>		2	

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Q	uestion	Answer	Mark	Guidance
	(d)	Any three from:	3	
		indicator species tell you how polluted the environment is $\checkmark$		ALLOW simple reference to pollution
		different species tolerate different levels/types of pollution $\checkmark$		
		changes in populations indicate changes in water quality $\checkmark$		ALLOW plants die if the water is polluted
		changes in populations prompt further chemical analysis/investigations into causes of change $\checkmark$		
		pondweed provide a habitat for other species / presence of pondweed indicates that other species may also be present $\checkmark$		
		Total	15	

Question		ion	Answer	Mark	Guidance
7	(a)		Complex formation✓DensityOptometryRedox✓Spectroscopy	2	
	(b)	(i)	Silver chromate $\checkmark$	1	
		(ii)	Red / brown ✓	1	
	(c)		Wear gloves / use goggles / wear a lab coat/PPE / use a fume cupboard $\checkmark$	1	IGNORE wear a mask ALLOW protective gear
	(d)		<b>Any one from</b> : Wash the contaminate off (using cold water) $\checkmark$ Contact supervisor / senior technician / H&S officer $\checkmark$ Fill in the accident book $\checkmark$	1	
	(e)	(i)	(Molar mass of AgNO <sub>3</sub> =) 169.9 √	2	<b>ALLOW</b> 170
			(2.125 / 169.9 =) 0.0125 (mol) ✓		ALLOW ECF using 2.125 ÷ their formula mass ALLOW answer to 2 s.f., 0.013
		(ii)	$(0.0125 / 0.25 =)$ $0.05 (mol dm^{-3}) \checkmark$	1	ALLOW ECF answer (e)(i) ÷ 0.25
	(f)	(i)	(No. moles Ag <sup>+</sup> in 15.5 cm <sup>3</sup> silver nitrate = 15.5 x 0.1/1000 =)	1	
			1.55 × 10 <sup>-3</sup> √		
		(ii)	(No. moles Cl <sup>-</sup> ions =) $1.55 \times 10^{-3} \checkmark$	1	ALLOW ECF using the same answer to (f)(i) ALLOW answer to 2 s.f. e.g. 0.0016
		(iii)	(Concentration of chloride in diluted sample = 1.55 x10 <sup>-3</sup> / 0.025 =) $0.062 \text{ (mol dm}^{-3}) \checkmark$	1	ALLOW ECF answer f(ii) ÷ 0.025

Question		ion	Answer	Mark	Guidance
		(iv)	(Concentration of chloride in undiluted tap water = $5 \times 0.062 =$ ) $0.31 \pmod{\text{dm}^3} $	1	ALLOW ECF answer f(iii) x 5
		(v)	(Yes, because) the concentration of chloride in the sample is more than <b>0.01</b> (mol dm <sup>3</sup> ) $\checkmark$	1	ALLOW ECF from f(iv). This mark is for a conclusion based on a comparison between answer to (iv) and 0.01
	(g)		Erichrome Black T $\checkmark$	2	
			EDTA✓		
			Total	16	

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